ideal mixture

A mixture of substances **B**, **C**, ... is called an ideal mixture when $a_{\rm B} = x_{\rm B}$, $a_{\rm C} = x_{\rm C}$, ... or $f_{\rm B} = 1$, $f_{\rm C} = 1$ where $a_{\rm B}$ and $f_{\rm B}$ are the relative activity and the activity coefficient, respectively, of a substance **B** in a liquid or solid mixture.

Source:

PAC, 1994, 66, 533 (Standard quantities in chemical thermodynamics. Fugacities, activities and equilibrium constants for pure and mixed phases (IUPAC Recommendations 1994)) on page 543