

amphiprotic (solvent)

Self-ionizing solvent possessing both characteristics of Brønsted acids and bases, for example H₂O and CH₃OH, in contrast to aprotic solvent.

Source:

PAC, 1994, 66, 1077 (*Glossary of terms used in physical organic chemistry (IUPAC Recommendations 1994)*) on page 1083

Orange Book, p. 30