

Brønsted base

A molecular entity capable of accepting a hydron (proton) from an acid (i.e. a 'hydron acceptor') or the corresponding chemical species. For example: OH^- , H_2O , CH_3CO_2^- , HSO_4^- , SO_4^{2-} , Cl^- .

See also: conjugate acid-base pair

Source:

PAC, 1994, 66, 1077 (*Glossary of terms used in physical organic chemistry (IUPAC Recommendations 1994)*) on page 1091