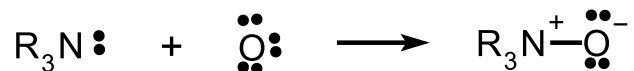


## dipolar bond

A bond formed (actually or conceptually) by coordination of two neutral moieties, the combination of which results in charge-separated structures, e.g.



The term is preferred to the obsolescent synonyms 'coordinate link', 'coordinate covalence', 'dative bond', 'semipolar bond'.

**Source:**

PAC, 1994, 66, 1077 (*Glossary of terms used in physical organic chemistry (IUPAC Recommendations 1994)*) on page 1106