

diradicals

Molecular species having two unpaired electrons, in which at least two different electronic states with different multiplicities [electron-paired (singlet state) or electron-unpaired (triplet state)] can be identified, e.g. $\text{H}_2\text{C}^{\cdot}-\text{CH}_2-\text{C}^{\cdot}\text{H}_2$ propane-1,3-diyl (trimethylene).

See also: biradical

Source:

PAC, 1995, 67, 1307 (*Glossary of class names of organic compounds and reactivity intermediates based on structure (IUPAC Recommendations 1995)*) on page 1333