

disproportionation

Also contains definition of: radical disproportionation

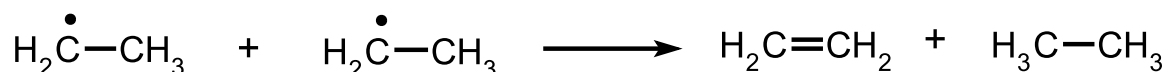
Synonym: dismutation

1.

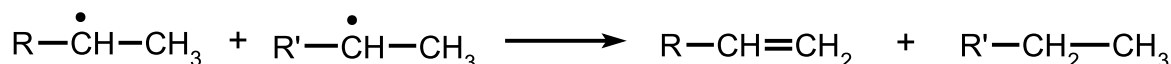
Any chemical reaction of the type
$$A + A \longrightarrow A' + A''$$
, where A, A' and A'' are different chemical species. For example:



The reverse of disproportionation is called comproportionation. A special case of disproportionation (or 'dismutation') is 'radical disproportionation', exemplified by:



Reactions of the more general type:



are also loosely described as radical disproportionations.

Source:

PAC, 1994, 66, 1077 (*Glossary of terms used in physical organic chemistry (IUPAC Recommendations 1994)*) on page 1107

The following somewhat more restricted usage of the term prevails in inorganic chemistry.

PAC, 1994, 66, 577 (*Definitions of terms relating to phase transitions of the solid state (IUPAC Recommendations 1994)*) on page 581

2. A reversible or irreversible transition in which species with the same oxidation state combine to yield one of higher oxidation state and one of lower oxidation

state. Example:
$$3 \text{Au}^+ \longrightarrow \text{Au}^{3+} + 2 \text{Au}$$
 The term also applies to an internal oxidation-reduction process as occurs, for example, among the iron atoms of

$$2 \text{Fe}^{4+} \longrightarrow \text{Fe}^{(4-\delta)+} + \text{Fe}^{(4+\delta)+}$$
 CaFeO₃, where , at Fe subarrays on lowering the temperature.

Source:

PAC, 1994, 66, 577 (*Definitions of terms relating to phase transitions of the solid state (IUPAC Recommendations 1994)*) on page 581