

exergonic (exoergic) reaction

This expression is often applied to a reaction for which the overall standard Gibbs energy change ΔG° is negative. Some workers use this term with reference to a negative value of ΔH° at the absolute zero of temperature.

Source:

PAC, 1996, 68, 149 (*A glossary of terms used in chemical kinetics, including reaction dynamics (IUPAC Recommendations 1996)*) on page 165