virial coefficients

Coefficients $(B, C, ...)$ in the virial equation of state for a real gas $p\,V_\mathrm{m} = R\,T\left(1 + \frac{B}{V_\mathrm{m}} + \frac{C}{V_\mathrm{m}^2} + ...\right)$ where p is the pressure, V_m the molar volume, R the gas constant and T the thermodynamic temperature.

Source:
Green Book, 2nd ed., p. 49

See also:
Purple Book, p. 58